

DIMETHYL ETHER AS A SUSTAINABLE FUEL FOR INTERNAL COMBUSTION ENGINES: OPPORTUNITIES AND OBSTACLES IN DECARBONIZATION

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ABSTRACT

The escalating concerns over climate change and dwindling fossil fuel reserves necessitate the urgent exploration of alternative, low-carbon fuels for internal combustion engines (ICEs). Dimethyl Ether (DME) has emerged as a promising candidate due to its favorable combustion properties, clean emission profile, and diverse feedstock options. This article, structured in the IMRaD format, comprehensively reviews the prospects and challenges associated with adopting DME as an alternative fuel in ICEs. It delves into DME's production pathways, thermophysical properties, its impact on engine performance and emissions, and the critical modifications required for its practical application. While DME offers significant advantages in reducing particulate matter and nitrogen oxides, challenges related to its low lubricity, bulk modulus, and the need for specialized fuel injection equipment and infrastructure must be addressed for its widespread adoption. This review synthesizes recent research, highlighting key advancements and persistent hurdles, to provide a holistic understanding of DME's role in the future of sustainable transportation.

Keywords: Dimethyl ether, sustainable fuels, internal combustion engines, decarbonization, alternative fuels, emission reduction, bio-DME, green transportation, combustion performance, clean energy.

INTRODUCTION

The global reliance on fossil fuels for energy, particularly in the transportation sector, has led to significant environmental degradation, including greenhouse gas (GHG) emissions and air pollution [7]. Internal combustion engines (ICEs), while central to modern transportation, are major contributors to these emissions, prompting an intensive search for cleaner, more sustainable fuel alternatives [5, 9]. As governments worldwide implement stricter emission regulations, the automotive industry faces immense pressure to develop and deploy low-carbon solutions [35].

Among the various alternative fuels being investigated, Dimethyl Ether (DME) stands out as a highly promising candidate for ICE applications, particularly in compression-ignition (CI) engines [3, 23, 29]. DME

(CH_3OCH_3) is a simple ether with a chemical structure similar to liquefied petroleum gas (LPG) but with the key advantage of containing oxygen, which promotes cleaner combustion [29, 31]. It possesses a high cetane number (55-60), making it an excellent fuel for CI engines, and its clean-burning nature leads to significantly reduced particulate matter (PM) and nitrogen oxide (NOx) emissions compared to conventional diesel fuel [3, 14, 26, 31].

The potential of DME extends beyond its combustion characteristics. It can be synthesized from a wide array of feedstocks, including natural gas, coal, biomass, and even municipal solid waste (MSW) and CO_2 , offering a pathway to diversify energy sources and reduce dependence on crude oil [4, 6, 10, 17, 18, 27, 29, 37]. This feedstock flexibility, coupled with its clean

combustion, positions DME as a vital component in the strategy for decarbonizing the transport sector and achieving energy security [8, 15].

Despite its inherent advantages, the widespread adoption of DME in ICEs is not without challenges. These include the need for specific fuel injection systems due to DME's low viscosity and lubricity, material compatibility issues with existing fuel lines, and the substantial investment required for production and distribution infrastructure [1, 15, 21]. This article aims to provide a comprehensive review of the current prospects and persistent challenges of utilizing dimethyl ether as a low-carbon alternative fuel in internal combustion engines, synthesizing insights from recent advancements and ongoing research.

2. METHODS

The assessment of Dimethyl Ether (DME) as a viable alternative fuel for Internal Combustion Engines (ICEs) necessitates a multi-faceted approach, encompassing its production pathways, inherent physicochemical properties, and the technical modifications required for its effective utilization in existing engine platforms. This section outlines the methodologies involved in understanding and evaluating DME's potential.

2.1. DME Production Pathways

DME can be synthesized through various routes, offering flexibility in feedstock sourcing and contributing to its sustainability profile [4, 29]. The primary methods include:

- **Indirect Synthesis (Two-step process):** This is the most common method, involving the initial synthesis of syngas (a mixture of CO, H₂, and CO₂) from diverse carbonaceous feedstocks such as natural gas, coal, biomass, or municipal solid waste (MSW) [2, 6, 17, 18]. The syngas is then converted into methanol, which subsequently undergoes catalytic dehydration to produce DME [4, 34]. This two-step process allows for feedstock flexibility and is commercially established [4].
- **Direct Synthesis (One-step process):** This method involves the direct conversion of syngas to DME in a single reactor, combining methanol synthesis and dehydration catalysts [4, 17]. This process is considered more efficient and environmentally friendly due to better heat integration and fewer purification steps [18]. Recent studies have also explored the production of DME from biogas, emphasizing its potential for bioeconomy paths and profitability [6]. Techno-economic analyses have also investigated DME production via the bi-reforming pathway for transportation fuel, evaluating its greenhouse gas emission impact [37]. The viability of MSW-derived

DME as a clean cooking fuel in regions like Kolkata, India, further highlights its diverse application potential [10].

2.2. Physicochemical Properties of DME

The unique physicochemical properties of DME dictate its behavior as an engine fuel and the necessary modifications to engine systems. Key properties include [3, 29]:

- **High Cetane Number:** With a cetane number typically ranging from 55 to 60, DME is an excellent fuel for compression-ignition (CI) engines, promoting smooth and rapid auto-ignition [3, 31]. This eliminates the need for ignition improvers often required for other alternative fuels.
- **Low Boiling Point:** DME has a boiling point of -24.8°C , meaning it is stored as a liquid under moderate pressure, similar to LPG [29]. This requires a pressurized fuel system.
- **Low Viscosity and Lubricity:** DME's very low viscosity and poor lubricity are significant challenges for conventional diesel fuel injection equipment (FIE), which relies on the fuel for lubrication [1, 21, 33]. This necessitates specialized FIE designs and materials.
- **Oxygen Content:** DME contains approximately 35% oxygen by mass [3, 29]. This inherent oxygenation contributes to more complete combustion and a significant reduction in particulate matter (PM) emissions [14, 26, 43].
- **Lower Heating Value:** DME has a lower heating value compared to diesel fuel, approximately 28.8 MJ/kg [3]. This means that a larger volume of DME is required to achieve the same energy output as diesel, which impacts fuel tank design and fuel consumption rates.
- **Miscibility:** DME is miscible with diesel in certain proportions, allowing for the potential use of DME/diesel blends, though the focus is often on pure DME engines due to its clean combustion benefits [43, 45].

2.3. Engine System Modifications and Experimental Approaches

Converting or designing ICEs to run on DME requires significant modifications, particularly in the fuel delivery and injection systems [21, 22, 23]. Experimental studies and numerical analyses are conducted to assess these modifications and their impact on engine performance and emissions:

- **Fuel Injection Equipment (FIE) Development:**

Due to DME's low viscosity and lubricity, specialized FIE is crucial. This involves developing new high-pressure pumps, injectors, and sealing materials that can withstand DME's properties [1, 19, 21, 33, 41]. Research focuses on optimizing nozzle hole diameters, injection pressure, and spray characteristics to ensure efficient atomization and mixing [19, 24, 25, 38, 41]. Studies have investigated the macroscopic spray characteristics and breakup performance of DME at high fuel temperatures and ambient conditions [25], and also the effects of impingement parameters on fuel-air mixture formation [38].

- **Combustion System Optimization:** Parameters such as injection timing, duration, and multiple injection strategies are optimized to achieve desired combustion phasing and reduce emissions [13, 14, 44]. Low-temperature combustion (LTC) strategies, such as Homogeneous Charge Compression Ignition (HCCI), are explored to further reduce NO_x and PM simultaneously [13, 15]. Numerical analyses are used to understand combustion and emissions formation in heavy-duty DME engines [28].
- **Emission Control Technologies:** While DME inherently reduces PM, strategies like Exhaust Gas Recirculation (EGR) and oxidation catalysts are investigated to further control NO_x and unregulated emissions (e.g., formaldehyde) [42, 47, 48]. Comparison studies have been conducted on emission characteristics of diesel- and dimethyl ether-originated particulate matters [39].
- **Material Compatibility:** Extensive testing is required to ensure that materials used in fuel tanks, lines, pumps, and injectors are compatible with DME and do not degrade over time [29].
- **Engine Control Unit (ECU) Calibration:** The ECU needs to be recalibrated to manage DME's different air-fuel ratio requirements, injection timing, and fuel quantity [22].
- **Experimental Test Beds:** Engine dynamometer testing is a standard method to evaluate engine performance (power, torque, brake thermal efficiency) and emissions (regulated and unregulated) under various operating conditions [20, 22, 46]. This includes assessing fuel consumption and optimizing engine parameters for minimal emissions [12].

These methodologies collectively contribute to understanding DME's potential, identifying technical barriers, and developing practical solutions for its integration into internal combustion engines.

3. RESULTS

Extensive research and experimental studies have

consistently highlighted the significant benefits and the technical hurdles associated with the use of Dimethyl Ether (DME) in internal combustion engines. The results reveal its strong potential as a clean alternative fuel, particularly in the realm of emissions reduction, while also pointing to areas requiring further development.

3.1. Engine Performance and Combustion Characteristics

DME's high cetane number (55-60) is a major advantage for compression-ignition (CI) engines, leading to rapid and stable combustion [3, 31]. Studies have shown that DME-fueled CI engines exhibit comparable or even superior thermal efficiency to diesel engines, especially after optimization [12, 20, 46]. For instance, Agarwal et al. (2023) demonstrated that optimized DME fuel injection systems could increase thermal efficiency while reducing emissions [1].

- **Combustion Improvements:** DME's oxygen content (35% by mass) promotes complete combustion, resulting in a shorter ignition delay and faster heat release rates compared to diesel [14, 24]. This can lead to quieter engine operation due to reduced combustion noise [3, 23]. Experiments on multi-cylinder DME engines with common-rail injection systems have shown robust combustion performance [44].
- **Power and Torque:** While DME has a lower volumetric energy density than diesel, optimized DME engines can achieve similar power and torque outputs by increasing the fuel injection quantity [22, 46]. Zhang et al. (2008) showed good performance of heavy-duty diesel engines fueled with DME [46]. The design and optimization of the fuel injection system play a crucial role in achieving competitive engine performance [21].
- **Fuel Consumption:** Due to its lower heating value, the mass-based fuel consumption of DME is higher than that of diesel for the same energy output [22]. However, the volumetric fuel consumption can vary depending on engine optimization and efficiency gains. Optimization studies focus on balancing fuel consumption with emission reduction [12].

3.2. Emission Reduction Potential

The most compelling results for DME as an alternative fuel are observed in its emission profile, particularly for pollutants associated with diesel combustion [3, 26, 40].

- **Particulate Matter (PM) and Soot:** DME combustion produces virtually no soot or particulate matter due to the absence of C-C bonds and the presence of oxygen in its molecular structure [3, 14, 26, 39]. This is a significant advantage over diesel fuel, which is a major contributor to PM emissions [9]. Experimental

studies confirm a near-zero PM emission level from DME engines [14, 20, 43, 47]. Wei et al. (2014) conducted a comparison study on the emission characteristics of diesel- and dimethyl ether-originated particulate matters, further solidifying this benefit [39].

- **Nitrogen Oxides (NOx):** The impact on NOx emissions is more nuanced. While higher combustion temperatures can lead to increased NOx, DME's rapid and complete combustion can, under optimized conditions, lead to reduced NOx compared to diesel [12, 26]. Strategies like Exhaust Gas Recirculation (EGR) and optimized injection timing have been effectively used to control NOx emissions in DME engines [13, 28, 42, 47]. Kim et al. (2011) showed reduction of exhaust emissions through HCCI combustion using advanced injection timing in a DME engine [13]. Putrasari and Lim (2021) specifically reviewed DME's ability to control NOx and PM emissions [26].
- **Unregulated Emissions:** Research also addresses unregulated emissions such as formaldehyde, which can be a concern with oxygenated fuels. Optimizing combustion parameters and utilizing oxidation catalysts can mitigate these emissions [47, 48].
- **Greenhouse Gas (GHG) Emissions:** The overall GHG emission reduction potential of DME depends heavily on its production pathway. Bio-DME, synthesized from renewable feedstocks like biomass or waste, offers a significant reduction in well-to-wheel GHG emissions compared to fossil diesel [6, 10, 37]. Even fossil-derived DME can offer some reduction compared to diesel due to its cleaner combustion [8].

3.3. Challenges and Required Modifications

Despite the promising emission results, several challenges need to be addressed for the widespread commercialization of DME as an engine fuel [15, 23].

- **Fuel Injection System (FIE) Adaptation:** This is one of the most critical challenges. DME's low viscosity (≈ 0.13 cSt at 20°C) and poor lubricity (around 1/50th of diesel) mean that conventional diesel FIEs are incompatible. DME can cause rapid wear of FIE components and leakage [1, 21, 33]. Therefore, specialized high-pressure fuel pumps, injectors (e.g., common-rail systems designed for DME), and sealing materials are essential [41]. Research is continuously focused on developing novel DME FIE, examining strategies and spray characteristics [19, 20]. The atomization characteristics of DME fuel injected through common-rail systems have been thoroughly investigated [33].
- **Material Compatibility:** DME's solvent properties can lead to degradation of certain elastomers

and plastics commonly found in conventional fuel systems [29]. All materials in contact with DME must be carefully selected to ensure durability and prevent leaks.

- **Fuel Storage and Supply Infrastructure:** DME requires a pressurized storage and distribution system, similar to LPG [29]. While this technology is mature, building a widespread DME fueling infrastructure would require substantial investment [8, 35]. The existing LPG infrastructure could potentially be adapted, but significant expansion would still be necessary.
- **Lower Energy Density:** The lower volumetric energy density of DME necessitates larger fuel tanks for equivalent driving ranges, which can be a practical constraint for vehicle design, especially in passenger cars [22].
- **Cold Start and Low Temperature Operation:** While DME has a low boiling point, ensuring efficient cold start and stable operation at very low ambient temperatures requires careful system design, including potentially heated fuel lines or specialized injection strategies [22].
- **Unregulated Emissions (Formaldehyde):** Although DME is cleaner in regulated emissions, there is a potential for increased formaldehyde emissions due to incomplete combustion, especially under certain operating conditions. Mitigation strategies, including exhaust after-treatment, are crucial [47, 48].

In summary, the results demonstrate that DME offers a compelling solution for reducing harmful emissions from ICEs, particularly PM. However, its unique physical properties necessitate significant engineering modifications, primarily in the fuel injection and delivery systems, and a robust infrastructure build-out to realize its full potential.

4. DISCUSSION

The findings from extensive research strongly support Dimethyl Ether (DME) as a promising low-carbon alternative fuel for internal combustion engines, particularly in the context of global efforts to decarbonize the transportation sector. The discussion here synthesizes the implications of the results, compares various approaches, and identifies future research directions.

4.1. Advantages and Performance Interpretation

The most significant advantage of DME, consistently highlighted in the results, is its ability to drastically reduce particulate matter (PM) and soot emissions, virtually eliminating them under optimized conditions

[3, 14, 26, 39]. This is a critical factor for improving air quality, especially in urban environments, and aligns perfectly with increasingly stringent emission regulations [7]. The inherent oxygen content of DME is the primary driver for this cleaner combustion, promoting complete oxidation of the fuel [3, 29]. This characteristic alone positions DME as a superior alternative to diesel from an air quality perspective.

Furthermore, DME's high cetane number (55-60) ensures excellent ignition characteristics in compression-ignition engines, allowing for a smooth and quiet combustion process without the need for ignition improvers [3, 31]. This contrasts favorably with other alternative fuels like natural gas, which often require spark ignition or pilot injection in CI engines. While DME's lower energy density implies higher mass-based fuel consumption, engine optimization through advanced injection strategies and combustion control can lead to comparable thermal efficiencies and power outputs to diesel engines, making it a viable energy carrier [12, 20, 22]. The continuous development of novel DME fuel injection equipment has been instrumental in enhancing both thermal efficiency and emission reduction [1].

The versatility of DME production pathways, including synthesis from biomass, natural gas, coal, and even waste streams like municipal solid waste and CO₂, offers a pathway to energy independence and a circular economy [4, 6, 10, 17, 18, 37]. This feedstock flexibility makes DME a highly sustainable option, capable of significantly reducing the carbon footprint on a well-to-wheel basis, especially when produced from renewable sources [37]. This aligns with global efforts to transition towards a bioeconomy [6].

4.2. Addressing the Challenges

Despite its impressive advantages, the widespread adoption of DME is contingent upon overcoming several notable challenges, primarily related to its unique physicochemical properties and the required infrastructure development.

- **Fuel Injection System (FIE) Development:** The low viscosity and poor lubricity of DME pose the most significant technical hurdle for engine integration [1, 21, 33]. Conventional diesel FIEs suffer from severe wear and leakage when exposed to DME. This necessitates the design and implementation of highly specialized FIE, including pumps, injectors, and sealing materials that can withstand DME's properties [19, 41]. Recent advancements in common-rail injection systems specifically for DME, focusing on spray characteristics and atomization, are critical steps forward [33, 44]. Continued research in material science for FIE components and advanced manufacturing techniques are essential to reduce costs and enhance durability.

- **Infrastructure Investment:** The need for a pressurized storage and distribution network, akin to LPG, requires substantial capital investment [8, 29, 35]. While existing LPG infrastructure offers a starting point, a dedicated DME infrastructure for widespread automotive use would be a considerable undertaking. Policy support, financial incentives, and collaborative efforts between governments and industry will be vital to accelerate this development. Lessons learned from the deployment of other alternative fuels can inform strategies for DME [8].

- **Energy Density and Tankage:** The lower volumetric energy density of DME means that larger fuel tanks are required to achieve a driving range equivalent to diesel vehicles [22]. This presents design challenges, particularly for passenger vehicles where space is at a premium. Research into more compact and efficient high-pressure storage solutions could mitigate this issue.

- **Unregulated Emissions:** While DME significantly reduces regulated pollutants, the potential for increased formaldehyde emissions needs continuous monitoring and mitigation strategies, such as optimized combustion strategies and effective after-treatment systems [47, 48]. Further research is needed to comprehensively characterize and control all unregulated emissions under various engine operating conditions.

4.3. Future Outlook and Research Directions

The future of DME as a mainstream fuel largely depends on continuous innovation and a supportive regulatory and economic environment.

- **Engine Optimization:** Further research should focus on advanced combustion concepts, such as optimized low-temperature combustion (LTC) and Homogeneous Charge Compression Ignition (HCCI) strategies tailored for DME, to simultaneously achieve ultra-low NO_x and PM emissions with high efficiency [13, 15]. The integration of advanced sensors and real-time control systems can enable adaptive engine operation for varying DME qualities and ambient conditions.

- **Multi-fuel and Blended Operations:** Exploring the potential of DME in dual-fuel applications (e.g., DME-diesel or DME-natural gas) or as a blending component could offer transitional pathways and leverage existing fuel infrastructures [45]. Studies on the spray characteristics of DME/diesel blended fuels are already underway [45].

- **Sustainable Production:** Continued emphasis on scaling up bio-DME production from diverse renewable feedstocks, coupled with carbon capture and utilization

technologies, will be crucial for maximizing its environmental benefits and ensuring long-term sustainability [6, 17, 37].

- **Policy and Standardization:** The development of clear policies, standards, and regulations for DME fuel quality, vehicle specifications, and refueling infrastructure is paramount for its commercialization [8].
- **Techno-economic Analysis:** Ongoing techno-economic analyses are necessary to assess the full life-cycle costs and benefits of DME, considering production, distribution, and engine modification costs, to ensure its economic viability compared to conventional and other alternative fuels [37].

In conclusion, DME presents a compelling case for decarbonizing internal combustion engines due to its clean combustion and diverse feedstock potential. While engineering challenges related to fuel injection and infrastructure remain, ongoing research and strategic investments are paving the way for DME to play a significant role in the future of sustainable transportation.

5. CONCLUSION

Dimethyl Ether (DME) stands as a highly promising, low-carbon alternative fuel with significant potential to reshape the landscape of internal combustion engines. Its inherent oxygen content and high cetane number facilitate remarkably clean combustion, virtually eliminating particulate matter and substantially reducing other harmful emissions such as nitrogen oxides when optimized. This clean-burning characteristic is a critical advantage in addressing escalating concerns about air quality and public health. Furthermore, the ability to synthesize DME from a wide array of feedstocks—ranging from natural gas and coal to renewable sources like biomass, municipal solid waste, and even CO₂—underscores its versatility and potential to contribute significantly to energy security and decarbonization efforts.

Despite these compelling advantages, the widespread adoption of DME necessitates overcoming several substantial challenges. The most prominent technical hurdle lies in adapting fuel injection equipment to accommodate DME's low viscosity and poor lubricity, requiring specialized designs and material selections for pumps, injectors, and sealing components. Additionally, the establishment of a robust and widespread infrastructure for DME production, storage, and distribution, which operates under moderate pressure, demands considerable investment and strategic planning. Material compatibility within existing engine systems also requires careful consideration and dedicated solutions.

Ongoing research and development efforts are steadily addressing these challenges, with notable advancements in novel fuel injection systems and a deeper understanding of DME combustion characteristics. As global environmental regulations become increasingly stringent and the demand for sustainable energy solutions intensifies, DME is poised to play an increasingly vital role. Continued investment in research, infrastructure development, and supportive policy frameworks will be crucial in realizing DME's full potential as a key enabler for cleaner, more sustainable internal combustion engines in the transition to a low-carbon future.

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